## ISOMACEDONIC ACID FROM THE ROOTS OF Glycyrrhiza echinata

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The neutral products from the acid hydrolysis of an aqueous extract of <u>Glycyrrhiza echinata</u> were obtained by the method described previously [1]. Chromatographic separation of the neutral products on  $Al_2O_3$  (inactivated) with chloroform elution yielded several individual substances. One of them, with the composition  $C_{31}H_{48}O_4$ , mp 222-225°C, M<sup>+</sup> 484 (mass spectrometry), IR spectrum, cm<sup>-1</sup>: 1725, 3300, 3540,  $R_f$  0.4 [inactivated  $Al_2O_3$ , petroleum ether-diethyl ether (1:2)], forms a diacetate with mp 229-230°C (IR spectrum: 1735 cm<sup>-1</sup>) and, on saponification, a dihydroxy acid with mp 285°C (IR spectrum: 1703, 3450 cm<sup>-1</sup>), the diacetate of which has mp 282°C (IR spectrum: 1710, 1740, 3300 cm<sup>-1</sup>). The UV spectra of the dihydroxy acid and of all the derivatives mentioned have  $\lambda_{max}$  281 nm (log  $\varepsilon$  3.5-3.7). All the derivatives were obtained by the usual methods. The properties described above permitted the conclusion that the substance isolated was the methyl ester of a triterpene dihydroxy acid containing a conjugated system of double bonds in one ring.

The oxidation of the diacetate of the methyl ester of the substance with mp 229-230°C with selenium dioxide in acetic acid gave a product with mp 236-237°C. UV spectrum:  $\lambda_{max}$  259, 250, 241 nm (log  $\epsilon$  4-4.15). The properties of this substance (UV and IR spectra, melting point) were identical with those of the diacetate of methyl macedonate and gave no depression in a mixed melting point [2]. Saponification yielded macedonic acid. Apparently, in this case migration of the double bonds under the influence of SeO<sub>2</sub> takes place in the same way as has been observed in the conversion of isomeristotropic acid into meristotropic acid [3]. Furthermore, the products of the oxidation of the methyl ester diacetate studied and of the diacetate of the methyl macedonate were found to contain a common oxidation product, probably a dienedione with mp 298°C (IR spectrum, cm<sup>-1</sup>: 1620, 1660, 1690, 1740).

Thus, bearing in mind the possible structure of macedonic acid, structure (I) may be proposed for the substance under investigation [4].



The presence in the roots of <u>A</u>. echinata of 3,19-dihydroxyolean-9,11:12,13-diene-17-carboxylic acid confirms the fact that in the false liquorices the triterpene compounds are found mainly in the form of homo- or heteroannular dienes.

## LITERATURE CITED

- 1. N. P. Kir'yalov and T. N. Naugol'naya, Zh. Obshch. Khim., 33, 22 (1963).
- 2. N. P. Kir'yalov, in: Questions of the Study and Use of Liquorice in the USSR [in Russian], Leningrad (1966), pp. 123-129.
- 3. N. P. Kir'yalov and G. S. Amirova, Khim. Prirodn. Soedin., 150 (1968).
- 4. N. P. Kir'yalov, Khim. Prirodn. Soedin., 448 (1969).

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